

Research Article

# Theoretical Properties Prediction of Diesel-Biodiesel-DEE Blend as a Fuel for C.I. Engine With Required Modifications for Optimum Performance

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## Abstract

The reduction in vehicular exhaust emissions with high engine performance are measure concern regarding engine design. To control exhaust emissions of internal combustion engine the governmental & environmental agencies implemented stringent emission norms. These stringent emission norms are very difficult to achieve through advance engine technologies. Also rapid depletion of petroleum based fuels with increasing prices is measure issue now-a-days. Hence the interest of researchers increases towards the development of renewable alternative fuels for I.C. engine the renewable fuels having capacity to reduce emissions than petroleum based fuel. Diethyl ether is one of the promising renewable fuels which reduce the exhaust harmful emissions with improved performance parameters. The DEE is promising alternative fuel for C.I. engine obtained from dehydration process of ethanol. Many researchers investigated the suitability of DEE as an alternative fuel for C.I. engine without any modifications with limited percentage of DEE in diesel & biodiesel. The aim of this research work is to improve the C.I. engine performance with required modifications for diesel-biodiesel-DEE blended fuel.

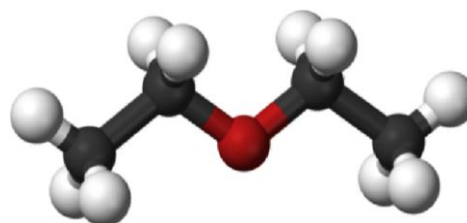
**Keywords:** Diethyl Ether, Alternative Fuel, Properties, Modifications

## 1. Introduction

As compared to petrol engine, diesel engines are widely used in worldwide as a prime mover because of its high thermal efficiency, high torque capacity, low HC & CO emission etc. So diesel engines cannot replace by any other prime mover in near future. Various technologies are being developed to face the problem of depleting resources of petroleum fuel & lower emissions like EGR technology, CRDI system, HCCI & PCCI combustion technology, dual fuel injection etc. Using these various engine & combustion technologies stringent emission norms cannot be meet. Biodiesel, ethanol, methanol, vegetable oils, LPG, CNG etc. are the various renewable alternative fuels are being used to meet the stringent emission norms & improve engine performance. These alternative fuels having some disadvantages like high viscosity, low cetane number, storage problem, low energy density, safety requirement & miscibility limit with diesel.

Ethanol is one of the promising high oxygen contained alternative fuel. But it has many obstacles to use in C.I engine because of low cetane number (Istvan *et al*, 2011), poor combustion characteristics & limited miscibility with diesel fuel. To overcome these

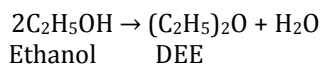
problems, ethanol can be converted easily into DEE through a dehydration process. DEE has several favorable properties for CI engines such as high cetane number (>125), low auto ignition temperature, high oxygen content, reasonable energy density for on-board storage, broad flammability limits, high miscibility with diesel fuel and renewable bio-fuel. DEE is a pungent, volatile, highly flammable liquid and widely used as a common solvent. It is the simplest ether expressed by its chemical formula  $\text{CH}_3\text{CH}_2\text{-O-CH}_2\text{CH}_3$ , consisting of two ethyl groups bonded to a central oxygen atom as shown in Fig. 1. (Patil & Thipse, 2015)



**Fig.1** Chemical Structure of DEE Molecule

The national renewable energy laboratory (NREL) conducted a process simulation exercise which showed that hydrous ethanol (ethanol diluted with water) can be converted to DEE. The main reactions is,

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The simulation also showed that the resulting liquid phase of water-ethanol/ ethanol-DEE could be separated in simple decanter separator. Ideally hydrous ethanol would be generated at the end of biomass ethanol production process so that the ether conversion process would take place before the final distillation in that process. The total process is simulated in fig.2.

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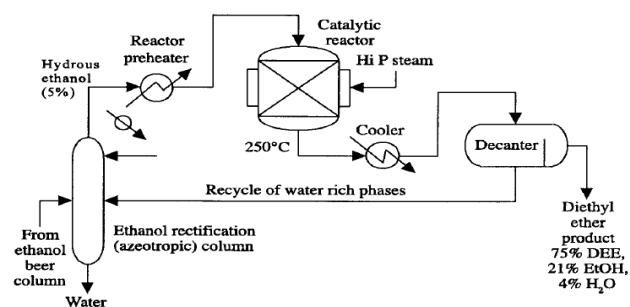


Fig.2 Production of DEE from Ethanol

## 2. Literature Review

The physical & chemical properties of DEE are good from emission & performance point of view in C.I. engine. Many researchers studied theoretically & experimentally the suitability of DEE as fuel for diesel engine. The experiment was carried out to investigate the effect of DEE with diesel-kerosene blend. It is found that optimum performance & emission levels achieved at 15% of DEE in diesel-kerosene blend. Even though DEE has high cetane number the ignition delay of diesel-DEE blends higher as compared to diesel. Also as the percentage of DEE to diesel-kerosene blend increases, the kinematic viscosity, density, calorific value of blend decreases & cetane number, oxygen content increased (Patil & Thipse, 2015). In the study of effect of DEE with neem oil biodiesel it is found that CO & smoke emission were reduced while HC & NO<sub>x</sub> emissions were increased for 5% DEE blended fuel. Also BSFC & brake thermal efficiency were improved for DEE-biodiesel blend. Up to 5% of DEE is promising technique (S. Sivalakshmi & T. Balusamy, 2013). In the experimental evaluation of performance & emission characteristics of diesel engine using diesel-DEE blend it is found that as the percentage of DEE in blend increases, the smoke, NO<sub>x</sub> & CO emissions goes on decreasing with increased HC emission. Regarding performance point of view addition of DEE to diesel

increases BSFC with constant brake thermal efficiency. Hence blending of DEE to diesel up to high blending ratio is advantageous from both emission & performance point of view (D. C. Rakopoulos *et al*, 2012). The effect of DEE & waste plastic pyrolysis oil on C.I. engine investigated experimentally. It is concluded that addition of DEE as an oxygenated fuel improves performance parameters & reduce tail pipe emission levels (J. Devaraj *et al*, 2015). The experimentally comparison between the effects of various percentage of DEE to biodiesel on combustion, performance & emission parameters of diesel engine is carried out to analyze the potential of DEE. It is reported that up to 20% DEE in biodiesel gives better stability with reduction in smoke & NO<sub>x</sub> emissions. Also HC & CO emissions were slightly higher as compared to biodiesel. Hence up to 10-20% of DEE as an oxygenated fuel can be added as an alternative fuel to biodiesel (S. Sudhakar & Dr. S. Shivprakasham, 2014).

From literature review it is found that up to 10-15% of DEE-diesel blend gives best results from emission & performance point of view. Above 15% of DEE in diesel results in poor performance or instable operation & power output of C.I. engine. It is due to high volatility, lower kinematic viscosity & density of DEE. This leads to vapor lock & increased leakage in fuel system. This can be resolved by addition of biodiesel to diesel-DEE blend (Patil & Thipse, 2015), (Istvan *et al*, 2011).

The main purpose of this investigation is to find out optimum blending ratio according to ASTM D975 standards. Also it is desired to find out modifications required in conventional diesel engine using diesel-biodiesel-DEE blend.

## 3. Attractive Properties of DEE

There are number of physical & chemical properties of fuels which having direct impact of performance & emission parameters of diesel engine. The various physiochemical properties of diesel, biodiesel & DEE compared in table 1.

### 3.1 Cetane Number

Cetane number is a primary indicator of easy in self-ignition of fuel. Cetane no is an important property associated with ignition delay. High cetane number shows the good auto ignition quality of compression ignition fuel with lower ignition delay. Cetane number of DEE is very high (125>) as compared to various compression ignition fuels.

### 3.2 Oxygen content

The addition of oxygen content additive to diesel is one of the promising techniques to reduce soot formation inside the combustion chamber. Oxygen content of DEE is high (21.6% by mass) as compared to diesel & diesel ethanol blend. Addition of oxygen to fuel results more complete combustion of fuel & reduces harmful toxic emission.

**Table 1** Physicochemical Properties of Diesel, Biodiesel & DEE

Properties	Diesel	Biodiesel	DEE
Chemical Structure	C <sub>10</sub> - H <sub>25</sub>	C <sub>12</sub> - H <sub>22</sub>	C <sub>2</sub> H <sub>5</sub> -O-C <sub>2</sub> H <sub>5</sub>
O <sub>2</sub> Content (mass %)	0	10-12	21.6
Cetane number	52	51	>125
Auto-ignition Temp.(°C)	250	-	160
Boiling Point at 1 atm. (°C)	146-374	182-337	34.6
Lower Calorific Value (MJ/kg)	43.26	39.83	33.9
A/F ratio	14.7	13.8	11.1
Liquid Density at NTP (kg/m <sup>3</sup> )	836	864	713
Viscosity of Liquid at NTP (cSt)	2.45	4.77	0.23

### 3.3 Auto ignition Temperature

The temperature at which a substance can be brought to flames without any external source like spark is called auto ignition temperature. Low auto-ignition temperature leads to good combustion due to this the NO<sub>x</sub> emission is reduced. Due to low auto-ignition temperature (160° c) of DEE it is also known as cold start aid fuel.

### 3.4 Density

Fuel density is mass per unit volume. As cetane number, viscosity & heating value are strongly related with density it has direct impact on performance & emissions parameters. The quality of atomization & combustion depends upon density (Zayed & Ahmed, 2014) (D. H. Qi *et al*, 2011). Change in fuel density affect the energy content of the fuel & alters the air fuel ratio. As higher density fuel produces more power it is important property from fuel economy point of view.

### 3.5 Viscosity

Viscosity is a measure of a liquids resistance to flow. The quality of spray, ease of starting the engine, the size of the particles and the quality of the fuel-air mixture in combustion chamber depends upon viscosity of fuel. The viscosity of DEE is very low as compared to other conventional fuel.

## 4. Predication of Properties of DEE Blends

There are various formulas to calculate the properties of the DEE blends. As the proportion of different constituent in blend changes its properties goes on changing. The 15% of biodiesel in diesel is best blend from performance point of view. For 15% biodiesel in diesel gives best BSFC & BTE improved by 1.17% (Shekar & Prabhakar, 2015). Hence DEE percentage is varied from 10 to 30% in B15 (85% diesel & 15% biodiesel) blend.

### 4.1 Cetane number

$$CN_{blend} = \sum_{i=1}^n CN_i \times X_i \quad (1)$$

Where, CN<sub>i</sub> is the cetane number of each constituent & X<sub>i</sub> is the % volume of each constituent in blend.

Oxygen Content:-

$$O_{blend} = \frac{\sum_{i=1}^n (\rho_i \times C_i \times X_i)}{\sum_{i=1}^n (\rho_i \times X_i)} \quad (2)$$

Where, ρ<sub>i</sub> is the density of each constituent, C<sub>i</sub> is the oxygen content of each constituent & X<sub>i</sub> is the % volume of each constituent in blend.

### 4.2 Density

$$\rho_{blend} = \sum_{i=1}^n (\rho_i \times X_i) \quad (3)$$

Where, ρ<sub>i</sub> is the density of each constituent & X<sub>i</sub> is the % volume of each constituent in blend.

### 4.3 Lower Calorific value

$$CV_{blend} = \frac{\sum_{i=1}^n (\rho_i \times X_i \times CV_i)}{\sum_{i=1}^n (\rho_i \times X_i)} \quad (4)$$

Where, CV<sub>i</sub> is the lower calorific value of each constituent, ρ<sub>i</sub> is the density of each constituent & X<sub>i</sub> is the % volume of each constituent in blend.

### 4.4 Kinematic Viscosity

$$\ln \mu_{blend} = \sum_{i=1}^n X_i \times \ln \mu_i \quad (5)$$

Where, μ<sub>i</sub> is the kinematic viscosity of each constituent & X<sub>i</sub> is the % volume fraction of each constituent in blend.

The table 2 shows the important properties of the blends of DEE with diesel & biodiesel. In this work properties are calculated for various percentage of diesel-biodiesel-DEE blends. From table D denotes for pure diesel, BD denotes for pure biodiesel & DEE denotes for DEE.

According to diesel fuel requirement of auto fuel vision and policy 2025 government of India report & ASTM D975 standard 35% diesel, 20% DEE & 45% biodiesel gives optimum blending ratio. For optimum blend ratio the as compared to pure diesel cetane number increased by 27.21%, oxygen content increased from 0 to 14.57% by mass which is advantageous from better combustion point of view.

**Table 2** Properties of diesel-biodiesel-DEE blends

Fuel	CN	O <sub>2</sub> Cont.	Density	Low. Cal. Value	Viscosity
Diesel	52	0	836	43.26	2.45
Biodiesel	51	12	864	39.83	4.77
Diethyl Ether	125	21.6	713	33.9	0.23
D75BD15DE10	59.2	3.7	827.9	41.91	2.13
D70BD15DE15	62.8	4.7	821.75	41.5	1.89
D65BD15DE20	66.5	5.7	815.6	41.07	1.68
D60BD15DE25	70.0	6.7	809.45	40.64	1.49
D55BD15DE30	73.8	7.7	803.3	40.2	1.33
D35BD40DE20	66.2	14.57	824	40.02	2.06

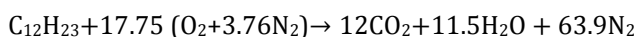
The decrease in density, lower calorific value & kinematic viscosity of the blend by 1.43%, 7.48%, & 15.91% respectively for optimum blend ratio as compared to pure diesel. The reduced viscosity is advantageous from atomization point of view.

### 5. Stoichiometric Air Fuel Ratios

For complete combustion of a fuel the required minimum amount of air is called stoichiometric air or also called theoretical air.

#### 5.1 Stoichiometric A/F for Diesel

The chemical formula for petroleum diesel is ranging from C<sub>10</sub>H<sub>20</sub> to C<sub>15</sub>H<sub>28</sub>. The average chemical formula for common diesel is C<sub>12</sub>H<sub>23</sub>. The combustion equation for diesel fuel is as,



The required amount air for combustion is 17.75 moles of oxygen & 17.75 x 3.76 moles of nitrogen, giving a total of 84.49 moles of air per mole of fuel. On mass basis the air fuel (AF) ratio for above combustion reaction is,

$$\text{AF} = \frac{\text{Mass of air}}{\text{Mass of fuel}}$$

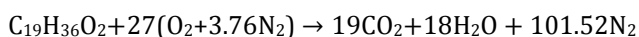
$$\text{AF} = \frac{84.49 \times 28.97}{12 \times 12 + 23}$$

$$\text{AF} = 14.65.$$

The air fuel ratio for diesel is 14.65:1

#### 5.2 Stoichiometric A/F for Biodiesel

There are two types of biodiesel that are generally used when calculating combustion equation C<sub>19</sub> & C<sub>20</sub> chain length biodiesel. The average chemical formula for biodiesel is C<sub>19</sub>H<sub>36</sub>O<sub>2</sub>. The combustion equation for biodiesel using C<sub>19</sub> chain length is as below,



The amount of combustion air is 27 moles of oxygen & 27 x 3.76 moles of nitrogen, giving a total of 128.52

moles of air per mole of fuel. On mass basis the air fuel (AF) ratio for above combustion reaction is,

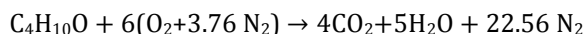
$$\text{AF} = \frac{128.52 \times 28.97}{19 \times 12 + 36 + 16 \times 2}$$

$$\text{AF} = 13.04$$

The air fuel ratio for biodiesel is 13:1

#### 5.3 Stoichiometric A/F for Diethyl Ether

The chemical formula for diethyl ether is C<sub>4</sub>H<sub>10</sub>O. The combustion equation for diethyl ether is,



The combustion air contains 6 moles of oxygen & 6 x 3.76 moles of nitrogen, giving a total of 28.38 moles of air per mole of fuel. On mass basis the air fuel (AF) ratio for above combustion reaction is,

$$\text{AF} = \frac{28.38 \times 28.97}{4 \times 12 + 10 + 16}$$

$$\text{AF} = 11.18$$

The air fuel ratio for DEE is 11.18:1. As compared to diesel & biodiesel the air fuel ratio for DEE is low due to its high oxygen content.

The table 3 shows the air fuel ratios for various blends of diesel, biodiesel & DEE. As the percentage of DEE in blend increases the air-fuel ratio goes on decreasing. As compared with pure diesel, the reduction in air fuel ratio for optimum blend is 8.33%.

**Table 3** Stoichiometric Air-Fuel Ratio for Diesel-Biodiesel-DEE Blends

Sr. No.	Fuel Blend	Stoichiometric A/F Ratio
1	Diesel (D100)	14.7:1
2	Biodiesel (BD100)	13.1:1
3	Diethyl Ether (DE100)	11.18:1
4	D75BD15DE10	13.83:1
5	D70BD15DE15	13.89:1
6	D65BD15DE20	13.79:1
7	D60BD15DE25	13.69:1
8	D55BD15DE30	13.58:1
9	D35BD45DE20	13.43:1

## 6. Modifications Required

### 6.1 Compression Ratio

The DEE has high cetane number as well as high latent heat of evaporation. It is expected that addition of DEE to diesel or biodiesel result in reduced ignition delay. But due to addition of DEE the latent heat of evaporation rate & increased ignition delay (Patil & Thipse, 2015), (J. Devaraj *et al*, 2014). Various researchers had studied effect of compression ratio on performance & emission. They reported that ignition delay reduced with increasing the compression ratio (Gnanamoorthi & Devaradjane, 2014), (Hariram & VageshShangar, 2015).

### 6.2 Injection Timing

Retarding or advancement in injection timing has direct impact on performance & emission parameters of engine. Hence the conventional diesel engines have to be modified for better performance & reduced emission with optimum injection timing.

### 6.3 Injection System

The conventional fuel systems are designed for injection of fuel only. The addition of DEE to diesel increases the volatility of the blend. Due to this the vapour percentage of fuel in injector is high. The conventional injection systems are not able to deliver both the liquid & vapour form of fuel. There is need to modify the injection system for delivering both liquid & vapour form fuel.

### 6.4 Intake system

DEE is very high oxygen contained fuel as compared to petroleum fuel. The addition of DEE reduces the air-fuel ratio of the blend. Hence intake system has to be modified for stoichiometric air fuel ratio.

## Conclusion

From theoretical study it is concluded that addition of DEE to diesel and biodiesel, increases the cetane number as well as oxygen content of the blend. Also it is observed that reduction in lower calorific value, density & kinematic viscosity. Due to high oxygen content, high cetane number & low viscosity, as the percentage of DEE increases, the performance parameters like BTE, BSFC etc. improved with lower emissions. But the addition of DEE to diesel and biodiesel is up to 15% gives better results. Hence with required modifications in conventional diesel engine the percentage of DEE to diesel and biodiesel can be increased for better performance & reduced emission parameters.

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